Reply to Office Action Dated: 13 January 2009

## **LISTING OF CLAIMS:**

- 1. (Previously Presented) A process for depositing a silver film on a substrate, comprising depositing silver on the substrate by chemical vapor deposition, in an oxygen atmosphere or in a hydrogen atmosphere, of a solution comprising a silver precursor, an amine and/or a nitrile, and a solvent, wherein:
  - the silver precursor is a silver carboxylate RCO<sub>2</sub>Ag in which R is a linear or branched alkyl radical that has 3 to 7 carbon atoms;
  - the concentration of the silver precursor in the solution is between 0.01 and 0.6 mol/l;
  - optionally the solvent has an evaporation temperature that is less than the decomposition temperature of the silver precursor; and
  - the percentage by volume of the amine and/or the nitrile in the solvent is more than 0.1%.
- 2. (Previously presented) The process according to claim 1, wherein the silver precursor is the silver pivalate (CH<sub>3</sub>)<sub>3</sub>-C-CO<sub>2</sub>Ag.
- 3. (Previously presented) The process according to claim 1, wherein the solvent is an organic compound that is liquid at ambient temperature and up to about 200°C under normal pressure conditions.
- 4. (Previously presented) The process according to claim 3, wherein the solvent is selected from the group consisting of mesitylene, cyclohexane, xylene, toluene and n-octane.
- 5. (Previously presented) The process according to claim 1, wherein the amine is a monoamine that is selected from the group consisting of n-hexylamine, isobutylamine, disec-butylamine, triethylamine, benzylamine, ethanolamine and disopropylamine.
- 6. (Previously presented) The process according to claim 1, wherein the amine is a polyamine.

Serial No. 10/550,459 Reply to Office Action Dated: 13 January 2009

7. (Previously presented) The process according to claim 1, wherein the nitrile is selected from the group consisting of acetonitrile, valeronitrile, benzonitrile and propionitrile.

- 8. (Previously presented) The process according to claim 1, wherein the substrate is formed by a material that is selected from the group consisting of superconductive high  $T_c$  materials, ceramics, thermoresistant polymers, glasses, MgO, LaAlO<sub>3</sub>, Ni, Si, AsGa, InP, SiC and SiGe.
- 9. (Previously presented) The process according to claim 1, wherein the temperature of the substrate on which silver is to be deposited is between 200 and 450°C.
- 10. (Cancelled)
- 11. (Previously presented) The process according to claim 1, wherein silver is deposited on the substrate in the presence of a cold plasma.